

MATERIALS RESEARCH SOCIETY OF SERBIA  
INSTITUTE OF TECHNICAL SCIENCES OF SASA

*Programme and the Book of Abstracts*

**NINETEENTH YOUNG RESEARCHERS' CONFERENCE  
MATERIALS SCIENCE AND ENGINEERING**

Belgrade, December 1-3, 2021



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**Materials Research Society of Serbia  
&  
Institute of Technical Sciences of SASA**

**2021**

Book title:

Nineteenth Young Researchers' Conference - Materials Science and Engineering:  
Program and the Book of Abstracts

Publisher:

Institute of Technical Sciences of SASA  
Knez Mihailova 35/IV, 11000 Belgrade, Serbia  
Tel: +381-11-2636994, 2185263, <http://www.itn.sanu.ac.rs>

Conference organizers:

Materials Research Society of Serbia, Belgrade, Serbia  
Institute of Technical Sciences of SASA, Belgrade, Serbia

Editor:

Dr. Smilja Marković

Technical Editor:

Aleksandra Stojičić

Cover page: Aleksandra Stojičić and Milica Ševkušić

Cover: Milica Ševkušić

Printing:

Gama digital centar  
Autoput No. 6, 11070 Belgrade, Serbia  
Tel: +381-11-6306992, 6306962  
<http://www.gdc.rs>

Publication year: 2021

Print-run:

120 copies

CIP - Каталогизacija у публикацији  
Народна библиотека Србије, Београд  
66.017/.018(048)

**YOUNG Researchers Conference Materials Sciences and Engineering (19 ; 2021 ; Beograd)**

Program ; and the Book of abstracts / Nineteenth Young Researchers' Conference Materials  
Science and Engineering, December 1-3, 2021, Belgrade, Serbia ; [organized by] Materials Research  
Society of Serbia & Institute of Technical Sciences of SASA ; [editor Smilja Marković]. - Belgrade :  
Institute of Technical Sciences of SASA, 2021 (Belgrade : Gama digital centar). - XVIII, 86 str. : ilustr.  
; 23 cm

Tiraž 120. - Registar.

ISBN 978-86-80321-36-3

а) Наука о материјалима -- Апстракти б) Технички материјали – Апстракти

COBISS.SR-ID 51231241

## Aim of the Conference

Main aim of the conference is to enable young researchers (post-graduate, master or doctoral student, or a PhD holder younger than 35) working in the field of materials science and engineering, to meet their colleagues and exchange experiences about their research.

## Topics

Biomaterials  
Environmental science  
Materials for high-technology applications  
Materials for new generation solar cells  
Nanostructured materials  
New synthesis and processing methods  
Theoretical modelling of materials

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### Results of the Conference

Beside printed «Program and the Book of Abstracts», which is disseminated to all conference participants, selected and awarded peer-reviewed papers will be published in journal “Tehnika – Novi Materijali”. The best presented papers, suggested by Session Chairpersons and selected by Awards Committee, will be proclaimed at the Closing Ceremony. Part of the award is free-of-charge conference fee at YUCOMAT 2022.

### Sponsors



**ANALYSIS**  
LABORATORY EQUIPMENT

### Acknowledgement

The editor and the publisher of the Book of abstracts are grateful to the Ministry of Education, Sciences and Technological Development of the Republic of Serbia for its financial support of this book and The Nineteenth Young Researchers' Conference - Materials Sciences and Engineering, held in Belgrade, Serbia.

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### Layered $\text{CaV}_2\text{O}_6$ as promising electrode material for multivalent storage

Tamara Petrović<sup>1</sup>, Miloš Milović<sup>2</sup>, Danica Bajuk-Bogdanović<sup>1</sup>, Milica Vujković<sup>1</sup>

<sup>1</sup>Faculty of Physical Chemistry, University of Belgrade, Belgrade, Serbia,

<sup>2</sup>Institute of Technical Sciences of SASA, Belgrade, Serbia

While the world is facing a higher demand for lithium, its limited resources associated with the high price, are becoming problematic. Other crucial drawbacks of Li-ion batteries are their toxicity and safety concerns. Therefore, researchers are oriented towards development of non-Li batteries based on eco-friendly and earth-abundant materials to overcome drawbacks of Li-ion technology. Alternative abundant metals and their ions such as Mg and Ca could be a good choice for rechargeable batteries in terms of cost and eco-friendliness.  $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$  ions could transfer two electrons per redox process which theoretically has a positive effect on battery performance. The materials upon which multivalent ions will intercalate with fast diffusion rate are hard to find. Metal vanadium oxide ( $\text{M}_x\text{V}_y\text{O}_z$ ) materials become promising materials for rechargeable batteries, so herein, a standard sol-gel combustion route was used for the preparation of the  $\text{CaV}_2\text{O}_6$  layered precursor. Two samples are synthesized from the vanadate precursor, the first when it was heated at 400 °C ( $\text{CaVO}$ ) and the second when  $\text{CaVO}$  was integrated with 10 wt % of sucrose under thermal treatment at 400 °C, in Ar atmosphere ( $\text{CaVO/C}$ ). Obtained  $\text{CaVO}$  and  $\text{CaVO/C}$  powders were thoroughly characterized by XRD, TG-DTA, FTIR, and Raman spectroscopy. The electrochemical performance of the obtained samples was evaluated for multivalent-ion storage in saturated aqueous electrolytic solutions of Mg ( $\text{NO}_3$ )<sub>2</sub> and Ca ( $\text{NO}_3$ )<sub>2</sub> by cyclic voltammetry and chronopotentiometry. For comparison, measurements were also done in saturated  $\text{LiNO}_3$ . Results indicated that  $\text{CaVO}$  can store more  $\text{Li}^+$  ions than  $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$  ions, but  $\text{CaVO}$  in  $\text{LiNO}_3$  shows a substantial loss of capacity upon cycling, which is not observed in the case of Mg ( $\text{NO}_3$ )<sub>2</sub> and Ca ( $\text{NO}_3$ )<sub>2</sub>. On the other hand,  $\text{CaVO/C}$  composite showed a significant improvement for Ca an Mg storage capacity, which exceeded capacity storage of  $\text{Li}^+$  ions. The high and stable discharge capacity of  $\text{CaVO/C}$ , amounting to 89.3 mA h g<sup>-1</sup> at 0.5 A g<sup>-1</sup>, was obtained in Ca ( $\text{NO}_3$ )<sub>2</sub>. Obtained results are promising and open novel directions regarding the use of  $\text{CaV}_2\text{O}_6$  for multivalent rechargeable batteries, especially for Ca-ion batteries.

*Acknowledgment:* This research was supported by the Science Fund of the Republic of Serbia, PROMIS, #6062667, HISUPERBAT.