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P.S.A.23.

HYDROTHERMAL SYNTHESIS OF LiFePO₄ POWDERS AS CATHODE MATERIAL FOR Li-ION BATTERIES

M. Jović, Z. Stojanović. Lj. Veselinović, D. Uskoković Institute of Technical Sciences of the Serbian Academy of Sciences and Arts, Belgrade, Serbia

Phospho-olivine LiFePO₄ have been intensively studied as lithium insertion cathode materials for next generation of Li-ion secondary batteries. LiFePO₄ has an interesting theoretical specific capacity of about 170 mAhg⁻¹, a good cycle stability and technically attractive flat voltage versus current profile of 3.45 V versus Li⁺/Li. A further advantage of this material, thanks to its stability, is the improved safety at high temperatures. In this work, LiFePO₄ was prepared by hydrothermal reaction starting from water solutions of LiOH, FeSO₄ and $\rm H_3PO_4$. After hydrothermal reaction the obtained powder was heat treated in reduction atmosphere to avoid oxidation of Fe²⁺ to Fe³⁺. Powder was additionally treated by high energy ball-milling. The structural and morphological properties of LiFePO₄ powder was characterized by X- ray diffraction and scanning electron microscopy. LiFePO₄/Li battery was characterized electrochemically by constant current charge-discharge cycling.

P.S.A.24.

IMPACT OF SOLVENT MIXTURE COMPOSITION AND ADDITIVE PRESENCE ON LiFePO $_4$ FORMATION IN WATER – ISO-PROPANOL SOLUTIONS AT ELEVATED TEMPERATURES AND PRESSURES

Z. Stojanović, M. Jović, D. Uskoković Institute of Technical Sciences of SASA, Belgrade, Serbia

High capacity storage of electric energy is technology of strategic importance for modern society. As lithium-ion batteries with their high capacity and generated power can improve electrical storage for electric vehicles and portable devices; they became a one of the major topics of research today. One of most perspective cathode material for lithium-ion batteries is LiFePO₄. Superior characteristics of this material are high Li storage capacity, stability, non-toxicity and low price, major disadvantage is low electronic conductivity. In this work we have investigated possibility to obtain pure LiFePO₄ nano-crystalline powders using water – iso-propanol mixture as reaction medium at temperatures in range of 180 to 220 °C and corresponding equilibrium pressures in stainless steel autoclave. Time was varied from 1 up to 24 h. In supplement, impact of different organic additives on morphology and size of particles is examined. Phase composition is determined by XRPD, morphology of particles by SEM and particle size distribution from light scattering measurements (LPSA). Electrochemical characterization of synthesized material is performed by constant current charge/discharge cycling.